Name_		Date	Hour	
² hy:	sical Science	Chapter 15-2	Properties of Matter	
1.	What is the difference between a physical and a chemical property?			
2.	Define physical property.			
3.	Give at least 4 examples or	f physical properties that a sample	e of matter might have.	
4.	Which property would we	measure if we used a balance or s	scale?	
5.	List a specific physical prop	perty of iron (many metals have th	nis property).	
6.	What is viscosity? How is	viscosity of maple syrup compare	to the viscosity of water?	
7.	Examine Figure 13, how w	ould you separate the mixture of	seeds?	
8.	Examine Figure 13, what p	hysical property does magnesium	have that sulfur does not?	
9.	Think critically: How could	d we separate a mixture of sand a	nd sugar?	
10.	What is a physical change?	?		

11.	Think critically: Does a change of state mean that a new substance is formed?
12.	How can we separate two different liquids that have been mixed together?
13.	What is a chemical property?
14.	Why are medicines often stored in dark colored bottles?
15.	What's a chemical change?
16.	What are 3 indicators that a chemical change is happening?
17.	Give an example of weathering that is would be considered a physical change.
18.	Give an example of weathering that would be considered a chemical change.
19.	What does the law of conservation of mass tell us?
20.	Think critically: If have a chemical reaction that says Element A + Element B → AB (a compound)
	If we have 10 grams of element A and 10 grams of element B, how many grams of the compound will be formed?