Name	Date	Hour

LaunchLAB

How do you know when a chemical change has occurred?

An indicator is a chemical that is added to the substances in a chemical reaction to show when change occurs.

Materials	Ammonia	Indicator Solution	Pipettes
	Water	Stir Rod	Effervescence Tablet
	Thermometer		

Procedure [Eye Safety, Clothing Protection, Irritant, Toxic, Handwashing]

- **1.** Put on your safety goggles; familiarize yourself with the materials at your table.
- Measure 20.0 mL of distilled water in a 25-mL graduated cylinder, and pour it into a 100-mL beaker. Using a pipette, add five drops of 0.1M ammonia to the water. WARNING: Ammonia vapors are extremely irritating.
- **3.** Stir 15 drops of **universal indicator** into the solution with a **stirring rod**. Observe the solution's color. Measure its temperature with a **thermometer**.
- **4.** Drop an effervescent tablet into a solution. Observe what happens. Record your observations, including any temperature change.
- **5.** Let the beaker sit for approximately 5 minutes. Record any additional observations on color and temperature.
- **6.** Empty your reaction beaker (100mL) into the sink, rinse it with water. Throw away any trash, push in your chairs.

Observations

Analysis

1. **Describe** any changes in the color or temperature of the solution.

2. Explain Was a gas produced? Is so, what did you observe to support this conclusion?

3. Analyze Did a physical change or a chemical change occur? Explain.