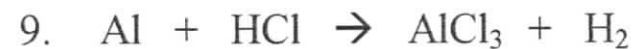
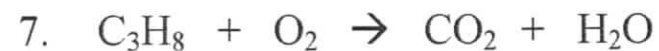
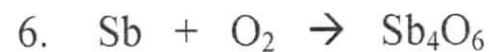
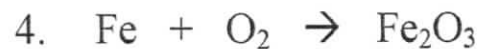


Name _____ Date _____ Hour _____

Balance the following reactions. Remember you can only add coefficients (Numbers in front of formulas).



11. $\text{N}_2 + \text{O}_2 \rightarrow \text{N}_2\text{O}$
12. $\text{Al} + \text{FeO} \rightarrow \text{Al}_2\text{O}_3 + \text{Fe}$
13. $\text{KI} + \text{Cl}_2 \rightarrow \text{KCl} + \text{I}_2$
14. $\text{Na} + \text{I}_2 \rightarrow \text{NaI}$
15. $\text{Cs} + \text{N}_2 \rightarrow \text{Cs}_3\text{N}_2$
16. $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$
17. $\text{K}_2\text{O} + \text{H}_2\text{O} \rightarrow \text{KOH}$
18. $\text{P}_4 + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$
19. $\text{Al} + \text{O}_2 \rightarrow \text{Al}_2\text{O}_3$
20. $\text{Zn} + \text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$